

Methyl 4-oxo-*r*-2,6-diphenylpiperidine-3-carboxylateN. Sampath,^a S. Aravindhan,^a
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Key indicators

Single-crystal X-ray study
T = 293 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
R factor = 0.045
wR factor = 0.136
Data-to-parameter ratio = 15.5For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The crystal structure of the title compound, $\text{C}_{19}\text{H}_{19}\text{NO}_3$, was determined because it is believed to possess medicinal properties. The piperidine ring assumes a chair conformation. The benzene-ring and methoxycarbonyl substituents are oriented equatorially. Whereas $\text{C}-\text{H}\cdots\text{O}$ intermolecular hydrogen bonds stabilize the crystal packing, the NH group does not form any hydrogen bond.

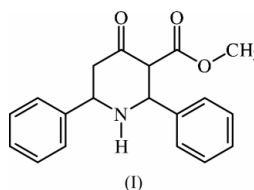
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Comment

Piperidine derivatives are used clinically to prevent post-operative vomiting, to speed up gastric emptying before anaesthesia, to facilitate radiological investigations and to correct a variety of disturbances of gastrointestinal functions (Robinson, 1973). Several 2,6-disubstituted piperidines are found to be useful as tranquilisers and some possess hypotensive activity (Kumar *et al.*, 1998), and a combination of stimulant and depressant effects on the central nervous system (Ganellin & Spickett, 1965), as well as bactericidal, fungicidal and herbicidal activities.



A *ZORTEP* plot (Zsolnai, 1998) of the title molecule, (I), is shown in Fig. 1. The piperidine ring adopts a chair conformation. Both benzene rings are attached to it in equatorial

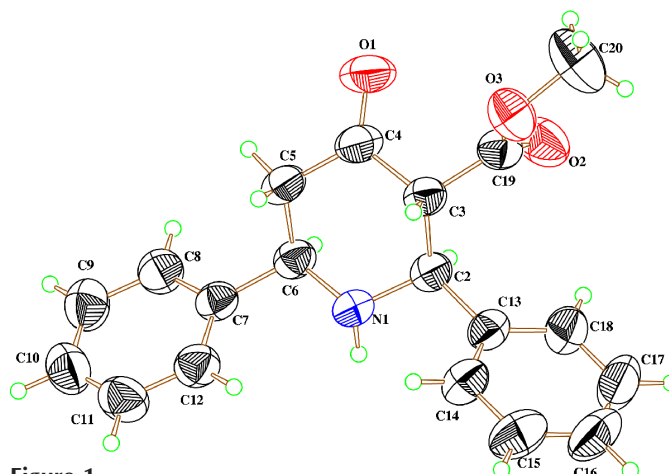


Figure 1
ZORTEP plot (Zsolnai, 1998) of the title molecule showing 50% probability displacement ellipsoids.

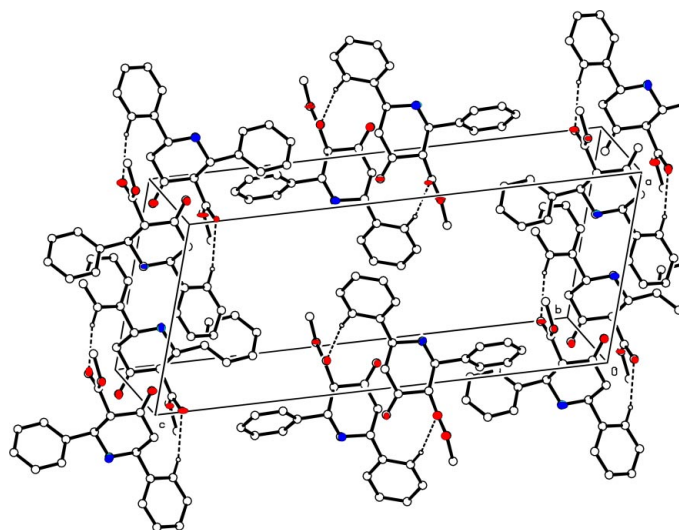


Figure 2
The packing of the title compound. The dashed lines show the hydrogen bonding.

positions. The dihedral angle subtended between the planes of the benzene rings is $32.1(4)^\circ$. The methoxycarbonyl group is attached to the piperidine ring in an equatorial position.

The packing of the molecules is shown in Fig. 2. $C-H \cdots O$ intermolecular hydrogen bonds stabilize the molecules in the crystal structure. The NH group does not form a hydrogen bond because it is shielded by the two adjacent benzene rings.

Experimental

The title compound was synthesized by a Mannich condensation reaction using benzaldehyde, methyl acetoacetate and ammonium acetate in a 2:1:1 ratio (Noller & Baliah, 1948) in 99% ethyl alcohol solution, refluxed for 1 h and then kept overnight. The colourless crystals obtained were recrystallized from ethanol by slow evaporation.

Crystal data

$C_{19}H_{19}NO_3$	$D_x = 1.226 \text{ Mg m}^{-3}$
$M_r = 309.35$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 3296 reflections
$a = 9.7563(13) \text{ \AA}$	$\theta = 1.8\text{--}26.0^\circ$
$b = 7.710(1) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$c = 22.810(3) \text{ \AA}$	$T = 293(2) \text{ K}$
$\beta = 102.348(4)^\circ$	Block, pale yellow
$V = 1676.1(4) \text{ \AA}^3$	$0.41 \times 0.20 \times 0.12 \text{ mm}$
$Z = 4$	

Data collection

Siemens SMART CCD area-detector diffractometer	2595 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.077$
Absorption correction: none	$\theta_{\text{max}} = 26.0^\circ$
12 559 measured reflections	$h = -12 \rightarrow 12$
3296 independent reflections	$k = -8 \rightarrow 9$
	$l = -28 \rightarrow 28$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0673P)^2 + 0.137P]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.136$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
3296 reflections	$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$
213 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.015 (2)

Table 1

Selected geometric parameters (\AA , $^\circ$).

C4—O1	1.2030 (17)	C19—O3	1.3321 (17)
C19—O2	1.1965 (17)		
C2—N1—C6—C7	176.64 (10)	C13—C2—C3—C4	-171.38 (11)
C2—N1—C6—C5	-63.66 (14)	C19—C3—C4—C5	169.90 (12)
C6—N1—C2—C13	-175.53 (10)	C2—C3—C4—C5	45.82 (16)
C6—N1—C2—C3	63.25 (14)	C3—C4—C5—C6	-47.47 (17)
N1—C2—C3—C19	-172.71 (10)	N1—C6—C5—C4	53.33 (15)
N1—C2—C3—C4	-51.02 (14)	C7—C6—C5—C4	174.75 (11)

Table 2

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C8-H8 \cdots O2^i$	0.930	2.49	3.386 (2)	163

Symmetry code: (i) $2 - x, -y, 1 - z$.

The H atom bonded to nitrogen was refined isotropically. H atoms bonded to carbon were positioned geometrically and refined using a riding model [$C-H = 0.93\text{--}0.98 \text{ \AA}$, and $U_{\text{iso}}(H) = 1.2U_{\text{eq}}(C)$ and $1.5U_{\text{eq}}(C_{\text{methyl}})$].

Data collection: *SMART* (Siemens, 2000); cell refinement: *SAINT* (Siemens, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1998) and *PLATON* (Spek, 1999); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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